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katelyn_matthews. chapter 11 vocab for Mrs. Brewer's AP class. Terms in this set (14) chemical equation. an expression representing a chemical reaction; the formulas of the reactants are connected by an arrow with the formulas for the products. skeleton equation. a chemical equation that does not indicate the relative amounts of reactants and

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products.

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Chem chapter 11 vocab Stoichiometry:
The study of quantitative relationships
between the amounts of reactants used
and products formed by a chemical

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reaction; is based on the law of conservation of mass. mole ratio: In a balanced equation, the ratio between the numbers of moles of any two substances. limiting reactant: A reactant that is totally consumed during a chemical reaction, limits the ...

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Chemistry 1405 Chapter 11 28. factor affecting atomic size - The number of protons in the nucleus (called the nuclear charge). The number of energy levels holding electrons (and the number of electrons in the outer energy level). 29. ion-an atom or molecule that has a net electrical charge. 30.

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PLAY. Photon. an elementary particle,
the quantum of light and all other forms
of electromagnetic radiation, and the
force carrier for the electromagnetic
force, even when static via virtual

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photons. Excited state.

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mattod43. Chemistry Chapter 11
Vocabulary. pressure. newton.

barometer. millimeters of mercury. the
amount of force exerted per unit area of
a surface. the SI unit for force; the force

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that will increase the speed.... an instrument that measures atmospheric pressure.

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Chapter 11 Vocabulary

1. amorphous solid - a solid whose molecular arrangement lacks a regular, long-range pattern
2. body-centered cubic - a cubic unit cell in which the lattice points occur at the corners and at the center
3. capillary action - the process by which a

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liquid rises in a tube because of a combination of adhesion to the walls of the tube and cohesion between liquid particles
4. coordination number - the number of adjacent atoms to which an atom is directly bonded
5 ...

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174051160: pressure: 1)the force applied to a unit area of surface 2)the amount of force exerted per unit area of a surface: 174051161: newton: 1)a unit of force equal to the force that imparts an 2)acceleration of 1 m/sec/sec to a mass of 1 kilogram the SI unit of force

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denature - There are two common

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meanings for this in chemistry. First, it can refer to any process used to make ethanol unfit for consumption (denatured alcohol). Second, denaturing can mean breaking down the three-dimensional structure of a molecule, such as a protein is denatured when exposed to heat.

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Chemistry Vocabulary - Definitions of Chemistry Terms

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Chapter 2 Chemistry of Life (Sect. 2-1

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and 2-2) Chap. 2.1

1. Atom – the basic unit of a chemical element
2. Nucleus – the positively charge central core of an atom
3. Electron – subatomic particle with a negative charge
4. Element – substance that cannot be chemically interconverted or broken down simpler
5. Isotope – forms of the same element containing equal number of protons but

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